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RESEARCH ARTICLE

Synthesis Characterization and Pharmacological Evaluation of Some Novel Trisubstituted Thiazole Derivatives

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ABSTRACT

Sets of trisubstituted thiazoles (*AR-11a-AR-53a*) were synthesized by reaction of substituted N-(morpholino/methoxyphenylpiperazin/methylpiperazine-1-yl-4-thiazol-5-carbonyl) benzamide (**3**) with (1-chloro-3-(2-methyl-4-nitro1H-imidazole-1-yl) propane-2-one), respectively. The (**3**) were synthesized by nucleophilic addition of benzoyl isothiocyanate /substituted benzoyl isothiocyanate and morpholino/methoxyphenylpiperazin/methylpiperazine-1-yl-4-thiazol-5-carbonyl in equimolar quantity at reflux temperature. The selected synthesized compounds from (*AR-11a-AR-53a*) were screened for their *in vivo* anti-inflammatory activity in carrageen in-induced rat hind paw oedema model at three graded doses employed at 10, 20 and 40 mg/kg body weight using Diclofenac sodium (10 mg/kg, 75 %) and ibuprofen (20 mg/kg, 74%) drug while other compounds were screened at single grad dose of 20 mg/kg. Among all the tested compounds *AR-21a*, *AR-31a*, *AR-12a* and *AR-42a* showed maximum antiinflammatory activity of 70% protection at 20 mg/kg and 56% protection at 40 mg/kg to inflamed paw, while other compounds showed Moderate to less protection at 10 mg/kg, 20 mg/kg & 40 mg/kg to inflamed paw.

KEYWORDS

Thiazoles, Nitroimidazole, Inflammation, Anti-Inflammatory Activity

INTRODUCTION

Inflammation is one of the important processes involved in the defense of an organism and it may be a normal and essential response to any noxious stimulus that threatens the host and may diverge from a localized to a generalized response, involving the whole organism.^{1,2} The stimulus may be external as with a burn, chemical irritation or mechanical trauma; or internal such as neoplasms, viruses or bacteria. Inflammation is the result of concentrated participation of a large number of vasoactive, chemolactic and proliferative factors at different

*Address for Correspondence: Aejaz Ahmed Abdul Rafique Head Department of Pharmaceutical Chemistry, Ali-Allana College of Pharmacy, Akkalkuwa Dist. Nandurbar, Maharashtra, India. E-Mail Id: aejazboraji@gmail.com stages and each type of stimulus provokes a characteristic prototype of response that represents a relatively minor variation on a theme. At a macroscopic level, the response usually is accompanied by the familiar clinical signs of erythema, edema, tenderness (hyperalgesia) and pain.³

Inflammatory bowel disease is a general term for a group of chronic inflammatory disorders of unknown etiology involving the gastrointestinal tract. Number of mediators such as prostaglandins, cytokines and reactive oxygen species are proposed to be involved in the pathogenesis of the diseases.^{4,5}

Multi-drug treatment of inflammatory conditions associated with microbial infections poses a

unique problem especially for patients with impaired liver or kidney functions. Therefore from the pharmacoeconomic and patient compliance points of view, the monotherapy with a drug having both anti-inflammatory and antimicrobial activities is highly desirable.⁶

Many attempts have been made to find a cure for inflammatory diseases and as a result, many nonsteroidal anti-inflammatory drugs such as phenylbutazone, oxyphenbutazone, diclofenac, ibuprofen, fenoprofen, indomethacin, benoxaprofen, benorylate, caprofen, tiopinac, ketoprofen and sulindac etc. are available in the market. These are more prevalent on chronic use and sometimes so severe that they can be fatal. This is an ugly facet of these invaluable therapeutic agents. This had geared the search for new and effective therapeutic agent, which could be used in the treatment of inflammation and pain.⁷

Various substituted thiazole have recently received attention of their diverse pharmacological properties, these includes anticancer, anti-inflammatory, anti-infective, antidiabetic, anti-HIV, anti-viral, anti-fungal, antioxidant, anesthetics, antibacterial, hypoglycemic, anticonvulsant activities.^{8,9,10}

Rajan S. Giri. et al.¹¹ reported a series of 2-(2,4di substituted-thiazole-5- yl)-3-aryl-3H-quinaz oline-4-one derivatives and evaluated for their inhibitory activity towards transcription factors TNFk B and AP-1 mediated transcriptional activation in a cell line based in vitro assay and for their anti-inflammatory activity in in vivo model of acute inflammation. Sherif Rostom AF et al.¹² synthesized antipyrine moieties attached to poly substituted thiazole and 2. 5disubstituted-1, 3, 4-thiadiazole through various linkages and evaluated for their antiinflammatory activity using the formalin-induced paw edema and the turpentine oil-induced granuloma pouch bioassays. Jagabandhu Das et al.¹³ reported a series of structurally novel aminothiazole based small molecule inhibitors of Itk, some compound identified as a potent and selective Itk inhibitor which reported to reduce lung inflammation in a mouse model of ovalbumin induced allergy/asthma. Taterao M. Potewar *et al.*¹⁴ were also reported a highly efficient and rapid synthesis of 2-amino-4arylthiazoles and 2-methyl-4-arylthiazole from a bromoketone and thiourea / thioamide was described using room temperature ionic liquid at ambient conditions. This methodology is successfully applied for a practical synthesis of an antiinflammatory agent, Fanetizole.

Sang Geon Kim et al.¹⁵ were also prepared the derivatives of thiazoles, including thiazole, benzothiazole and benzothiadiazole as intermediates in the production of a number of therapeutic agents. Benzothiazole derivatives anti-inflammatory exhibit analgesic and activities, possibly through the mechanism of 5lipoxygenase and thromboxane synthetase inhibition. Other benzothiazole compounds have gastric antisecretory activity or an anthelmintic effect. Shaohua Chang et al.¹⁶ designed a new series of 2-substitute d thiazole carboxamides were identified as potent pain inhibitors against all three isoforms of Akt (Akt1, Akt2 and Akt3) by systematic optimization of weak screening hit N -(1-amino-3-phenylpropan-2-yl)-2-phenylthiazole-5-carboxamide. The inhibitors might serve as lead compounds for further development of novel effective anticancer agents.

Moreover Jin-Hun Park *et al*.¹⁷ designed a series of new imidazo [2, 1- b] thiazole derivatives and *in vitro* antiproliferative activities against A375P human melanoma cell line and NCI-60 cell line panel were tested. Some compounds showed superior potency against A375P to sorafenib. Cristina Mariana *et al*.¹⁸ designed and reported series of novel acyl-hydrazones bearing 2-arylthiazole moiety by the condensation between derivatives of 4-[2-(4 -methyl-2phenyl-thiazole-5-yl)-2-oxo-ethoxy]-benzaldehyde and 2, 3 or 4-(2-aryl-thiazol-4-ylmethoxy)-benzaldehyde,

respectively and different carboxylic acid hydrazides. These compounds were tested *in vivo* for their anti-inflammatory activity, in an acute experimental inflammation.

Prakash K *et al.*¹⁹ synthesized series of novel 4aryl/chloroalkyl-2-(2, 3, 5-trichlorophenyl)-1, 3thiazoles by condensing 2, 3, 5trichlorobenzenecarbothioamide with phenacyl bromide/dichloroacetone. 2, 3, 5trichlorobenzaldehyde thiosemicarbazone on treatment with phenacyl bromide afforded 4-aryl-2-(2, 3, 5-trichlorophenylidenehydrazino)-1, 3thiazoles in good yield. These compounds were also screened for their antibacterial and antifungal activities.

Based on through envisaged and importance the emergence of challenging diseases associated with inflammatory conditions leads to development of novel series of substituted thiazole compound. Development was made taking into account of recent approaches in synthesis of thiazole containing drug proposed to act as excellent agent with multiple pharmacological response.

MATERIAL AND METHODS

Chemistry

All reagents and solvents were used as obtained from the supplier or recrystallized/ redistilled as necessary. Thin layer chromatography was performed on microscopic slides coated with silica gel G and and toluene: methanol as a mobile phase. The spots were visualized by normal TLC and exposure to iodine vapour. Melting points were recorded on open capillary melting point apparatus and were uncorrected. IR spectra were recorded in KBr on SHIMADZU Fourier Transform Infrared 8400 S spectrophotometer. Mass spectra were recorded on Electron impact (EI) on a Jeol JMS-D-300 spectrometer with the ionization potential of 70 eV. Nuclear Magnetic Resonance spectra (¹H NMR) were recorded in DMSO-d6 on Bruker advance at 400 MHz using Tetramethyl silane

(TMS) as internal standard and the chemical shift (δ) were reported in parts per million (ppm). Elemental analysis data were determined using a Carlo-Erba 1108 instrument or Elementar's Vario EL III micro-analyzer.

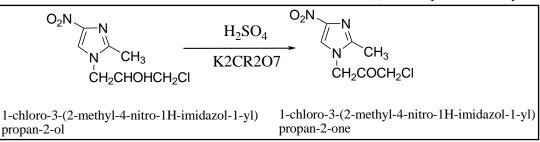
Synthesis of novel (1-chloro-3-(2-methyl-4nitro1h-imidazole-1-yl) propane-2-on)²⁰

20 g (0.09 mol) of (1-chloro-3-(2-methyl-4nitro1H-imidazole-1-yl) propane-2-ol) and 14 g (0.045mol) of potassium dichromate were placed in conical flask. To the reaction flask sulfuric acid was added dropwise at reduced temperature (15-20 °C). Once the mixture turned greenish to greenish black, stirring was started for 20 hour. The reaction container was removed and reaction mixture was transferred in to 200 mL of water. Reaction mixture was set aside for five days to obtained solid crystals of oxidized form (1chloro-3-(2-methyl-4-nitro1H-imidazole-1-yl)

propane-2-ol). Filtered& washed with water & dried at 60-70 °C in an oven. The yield of solid was obtained 5 g. The melting point was recorded 151-152 °C.²¹ Percentage Yield: 46 %, Melting point: 122 °C TLC: Toluene: Methanol (8.5: 1.5 Rf Value: 7.6)

General Method for Synthesis of Compounds (*AR-11a-AR-53a*)

As shown in Scheme II, isothiocyanate (2) were obtained by stirring ammonium thiocyanate (0.1379 mol) in 100 mL acetone at room temperature with benzoyl chloride/substituted benzoyl chloride (0.1263 mol) for 15-25 minutes followed by refluxing the reaction mixture for 15 min. Substituted N- (Morpholino/4-methoxyphenylpiperazinyl/N-methylpiperazinyl-4-thioyl) benzamide (3) were synthesized by

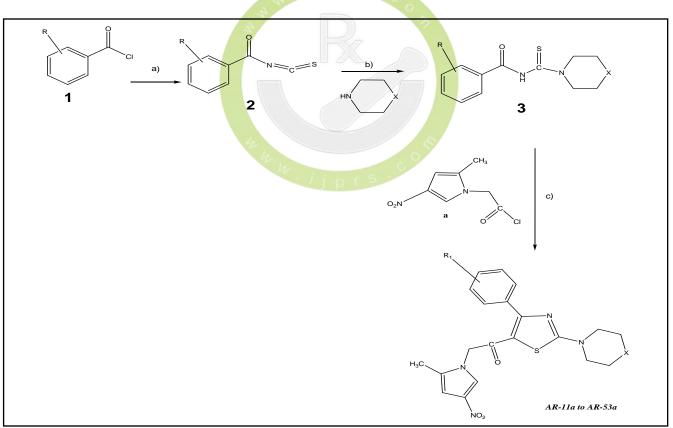


Scheme I: Synthesis of (1-chloro-3-(2-methyl-4-nitro1H-imidazole-1-yl) propane-2-on) Reagents and conditions: a) Stirred for 20 hr (1-chloro-3-(2-methyl-4-nitro1H-imidazole-1-yl) propane-2-ol) $K_2cr_2O_7, H_2SO_4$

nucleophilic addition of benzoyl isothiocyanate /substituted benzoyl isothiocyanate and morpholine/4-methoxyphenylpiperazine/4methylpiperazine in equimolar quantity at reflux temperature. The compounds (AR-11a-AR-53a) were synthesized as per procedure reported by Reji et al^{21.} In brief the 1-chloro-3-(2-methyl-4nitro1H-imidazole-1-yl) propane-2-on (a) was added in equimolar quantity to a solution of adduct (substituted N-(Morpholino/4-methoxyphenylpiperazinyl/N-methylpiperazinyl-4-thioyl) benzamide (3) in 10 mL acetonitrile. The reaction mixture was heated on a water bath at 100°C for 5 hr. The reaction mixture was cool and filtered. To the filtrate a pinch of sodium bicarbonate was added. Yellow precipitates thus obtained were filtered, wash with water, air dried and purified by recrystallization corresponding to the (AR-11a-AR-53a) and was characterized as per the analytical data.^{22,23,24}

AR-11a: 2-(2-methyl-4-nitro-1H-imidazol-1-yl)-1-(2-morpholino-4-phenylthiazol-5-yl) ethanone

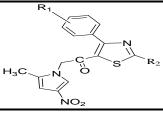
IR (KBr,Cm¹): 3150 (aromatic stretching), 2820 (-CH₃ stretching), 1626 (C=O stretching), 1444 $(NO_2-$ (C=C)stretching, aromatic). 1392 stretching), 1305 (N-C of aromatic), 908 (-C=C out of plane for benzene). 1120 (C-O-C stretching of morpholine).,¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.4 (t, 4H, morpholine N-CH₂), 3.3-3.6 (t, 4H, morpholine O-CH₂), 2.42 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 4.8 (s, 2H, -CH₂-C=O), 7.2-7.5-8 (m, 5H, aromatic proton at 4th position), 7.6 (s, 1H, aromatic proton at 5th position of nitroimidazole), m/z., 413.20, 414 (M+1) Anal for MS: C₁₉H₁₉N₅O₄S: Calculated C, 55.19; H, 4.63; N, 16.44 Found: C, 54.15; H, 4.43; N, 16.74;



Scheme II: Synthesis of 3-(4-(substituted) phenyl-2-morpholine/methoxyphenylpiperazin/methylpiperazine -1-yl-4-thiazol-5-carbonyl). Reagents and conditions: a) NH4SCN, acetone, reflux 15-25 min; b) reflux for 20 min, pour reaction mixture to crushed ice; c) acetonitrile, stir at 100°C for 5 h, pour to

crush

Table 1: Physicochemical properties of synthesized compound



Compound	R 1	R ₂	Melting Point	Rf Value	% Yield
AR-11-a	Benzoyl chloride	H-N_O	178-180	0.63	62%
AR-21-a	<i>p</i> -Chlorobenzoyl chloride	H-N_O	196-196	0.64	58%
AR-31-a	o-Chlorobenzoyl chloride	H-N_O	152-154	0.64	71%
AR-41-a	<i>m</i> -Chlorobenzoyl chloride	H-NO	154-158	0.64	62%
AR-51-a	Furoyl chloride	H-NO	218-220	0.69	68%
AR-12-a	Benzoyl chloride	H-N_N-OCH3	208-210	0.54	49%
AR-22-a	<i>p</i> -Chlorobenzo <mark>yl</mark> chloride	H-N_N-OCH3	170-172	0.56	48%
AR-32-a	<i>o</i> -Chlorobenzoyl chloride	H-N N-OCH3	177-179	0.55	64%
AR-42-a	<i>m</i> -Chlorobenzoyl chloride	H-N_N-OCH3	172-174	0.57	60%
AR-52-a	Furoyl chloride	H-N_N-OCH3	196-198	0.62	82%
AR-13-a	Benzoyl chloride	H-N_N-CH ₃	156-158	0.63	55%
AR-23-a	<i>p</i> -Chlorobenzoyl chloride	H-N_N-CH ₃	154-156	0.62	61%
AR-33-a	<i>o</i> -Chlorobenzoyl chloride	H-N_N-CH ₃	162-163	0.58	55%
AR-43-a	<i>m</i> -Chlorobenzoyl chloride	H-N_N-CH ₃	168-170	0.64	70%
AR-53-a	Furoyl chloride	H-N_N-CH ₃	158-160	0.49	61%

AR-21a: 1-(4-(4-chlorophenyl)-2-morpholinothiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1yl) ethanone

IR (KBr, Cm¹): 3130 (aromatic stretching), 2880 (-CH₃ stretching), 1660 (C=O stretching), 1510-1461 (C=C stretching, aromatic), 1450-1392 (NO₂- stretching), 1308 (N-C of aromatic), 827-914 (-C=C out of plane for mono substituted (C-O-C benzene), 1119 stretching of morpholine), 670 (-C-Cl Bending)¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.47 (t, 4H. morpholine N-CH₂), 3.45-3.76 (t, 4H. morpholine O-CH₂) 3.32 (s, 3H proton at 2nd position of nitroimidazole -CH₃) 5.40 (s, 2H, -CH₂-C=O), 7.35-7.59 (m, 4H, aromatic proton at 4th position), 8.1 (s, 1H, aromatic proton at 5th position of nitroimidazole) MS: m/z., 447, 448 (M+1)., Anal for C₁₉H₁₈ClN₅O₄S: Calculated: C, 50.95; H, 4.05; N, 15.64; Found: C, 50.65; H, 4.15; N, 14.65

AR-31a: 1-(4-(2-chlorophenyl)-2-morpholinothiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1yl) ethanone

IR (KBr, Cm¹): 3137 (aromatic stretching), 2890 (-CH₃ stretching), 1640 (C=O stretching), 1531-1446 (C=C stretching, aromatic), 1494-1392 (NO₂- stretching), 1308 (N-C of aromatic), 907(-C=C out of plane for mono substituted benzene). 1144-1120 (C-O-C stretching of morpholine), 650 (-C-Cl Bending)¹H NMR:(400 MHz, 2.29-2.56 DMSO-d6) (ppm): 4H. δ (t. morpholine N-CH₂), 3.45-3.76 (t, 4H. morpholine O-CH₂), 3.32 (s, 3H proton at 2nd position of nitroimidazole -CH₃) 5.42 (s, 2H, -CH₂-C=O), 7.32-7.39 (m, 4H, aromatic proton at 4th position), 7.98 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z.. 447.08.20, 448 (M+1)., Anal for C₁₉H₁₈ClN₅O₄S: Calculated C, 50.95; H, 4.05; Cl, 7.92; N, 15.64; Found: C, 50.55; H, 4.05; N, 15.44.

AR-41a: 1-(4-(3-chlorophenyl)-2-morpholinothiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1yl) ethanone

IR (KBr, Cm^1): 3204 (aromatic stretching), 2904 (-CH₃ stretching), 1704 (C=O stretching), 1530-1463 (C=C stretching, aromatic), 1469 (NO₂- stretching), 1342-1290 (N-C of aromatic), 836 (-C=C out of plane for mono substituted benzene), 1112 (C-O-C stretching of morpholine), 668 (-C-Cl Bending) ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.46 (t, 4H, morpholine N-CH₂), 3.45-3.76 (t, 4H, morpholine O-CH₂), 3.32 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 5.40 (s, 2H, –CH₂-C=O), 7.36-7.99 (m, 4H, aromatic proton at 4th position), 8.02 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 447.08, 448 (M+1)., Anal for C₁₉H₁₈ClN₅O₄S: Calculated: C, 50.95; H, 4.05; Cl, 7.92; N, 15.64 Found: C, 49.85; H, 4.05; N, 15.54

AR-51a: 1-(4-(furan-2-yl)-2-morpholinothiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 2979 (-CH₃ stretching), 1660 (C=O stretching), 1469-1446 (C=C stretching, aromatic), 1329 (NO₂- stretching), 1241 (N-C of aromatic), 1115 (C-O-C stretching of morpholine), 1018 (C-O-C stretching of furan) ¹H NMR:(400 MHz, DMSO-d6)δ (ppm): 2.29 (t, 4H, morpholine N-CH₂), 3.52-3.381 (t, 4H, morpholine O-CH₂), 3.32 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 5.4 (s, 2H, – CH₂-C=O), 7.32-7.39 (m, 3H, aromatic proton of Furoyl at 4th position), 7.98 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 403.1, 404 (M+1)., Anal for C₁₇H₁₇N₅O₅S: Calculated: C, 50.61; H, 4.25; N, 17.36 Found: C, 50.51; H, 4.15; N, 16.38

AR-12a: 1-(2-(4-(4-methoxyphenyl) piperazin-1-yl)-4-phenylthiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 3100 (aromatic stretching), 2800 (-CH₃ stretching), 1630 (C=O stretching), 1546 (C=C)stretching, aromatic), 15303 $(NO_2$ stretching), 1246-1226 (-OCH₃- stretching), 1332 (N-C of aromatic), 833 (-C=C out of plane for mono substituted benzene), 1153 (C-N-C stretching of phenyl piperazine) ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.16 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 3.01-3.31 (m, 8H, piperazine), 3.70 (s, 3H, O-CH₃), 5.47 (s, 2H, -CH₂-C=O), 6.8-6.95 (m, 4H, aromatic proton of 4-methoxyphenyl) 7.40-7.65 (m, 5H,

aromatic proton at 4th position), 8.0 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 518, 519 (M+1) , Anal for $C_{26}H_{26}N_6O_4S$: Calculated: C, 60.22; H, 5.05; N, 16.21; Found: C, 59.95; H, 5.05; N, 16.12;

AR-22a:1-(4-(4-chlorophenyl)-2-(4-(4-methoxyphenyl)piperazin-1-yl)thiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl)ethanone

IR (KBr,Cm¹).,3100 (aromatic stretching), 2800 (-CH3stretching), 1750 (C=O stretching), 1592 (C=C stretching, aromatic), 1546-1502 (NO₂stretching), 1270-1253 (-OCH₃- stretching), 1310 (N-C of aromatic), 909-831 (-C=C out of plane for mono substituted benzene), 1114-1013 (C-N-C stretching of phenyl piperazine), 750 (-C-Cl Bending) ., ¹H NMR:(400 MHz, DMSO-d6)δ (ppm): 2.42 (s, 3H proton at 2nd position of nitroimidazole –CH3) 3.33-3.65 (m, 8H, piperazine), 4.21 (s, 3H, O-CH₃), 4.91 (s, 2H, -CH₂-C=O), 7.40-7.52 (m, 4H, aromatic proton of 4-methoxyphenyl) 7.77-7.811 (m, 4H, aromatic proton at 4th position), 7.15 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 552.13, 553 (M+1), Anal for C₂₆H₂₅ClN₆O₄S: Calculated: C, 56.47; H, 4.56; Cl, 6.41; N, 15.20; Found: C, 55.47; H, 4.16; Cl, 6.41; N, 15.30;

AR-32a: 1-(4-(2-chlorophenyl)-2-(4-(4methoxyphenyl) piperazin-1-yl) thiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 3140 (aromatic stretching), 2900-2830 (-CH₃ stretching), 1690-1700 (C=O stretching), 1529 (C=C stretching, aromatic), 1502-1449 (NO₂- stretching), 1269-1248 (-OCH₃- stretching), 1310 (N-C of aromatic), 860-820 (-C=C out of plane for mono substituted benzene), 1166-1030 (C-N-C stretching of phenyl piperazine), 748 (-C-Cl Bending), ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.37 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 3.08-3.87 (m, 8H, piperazine), 4.28 (s, 3H, O-CH₃), 4.90 (s, 2H, -CH₂-C=O), 6.79-6.94 (m, 4H, aromatic proton of 4-methoxyphenyl) 7.3-7.54 (m, 4H, aromatic proton at 4th position), 8.21 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 553, 554 (M+1)., Anal for C₂₆H₂₅ClN₆O₄S: Calculated: C, 56.47; H, 4.56; Cl, 6.41; N, 15.20 Found: C, 55.47; H, 4.46; N, 15.20;

AR-42a: 1-(4-(3-chlorophenyl)-2-(4-(4methoxyphenyl) piperazin-1-yl) thiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 3120 (aromatic stretching), 2830 (-CH₃ stretching), 1700 (C=O stretching), 1529 $(NO_2-$ (C=C stretching, aromatic), 1502 stretching), 1269 (-OCH₃- stretching), 1310 (N-C of aromatic), 860-820 (-C=C out of plane for substituted benzene), 1182 (C-N-C mono stretching of phenyl piperazine), 682 (-C-Cl Bending)., ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.42 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 3.10-3.38 (m. 8H. piperazine), 3.70 (s, 3H, O-CH₃), 5.48 (s, 2H, -CH₂-C=O), 6.7-6.95 (m, 4H, aromatic proton of 4-methoxyphenyl) 7.48-7.65 (m, 4H, aromatic proton at 4th position), 7.21 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 553, 554 (M+1)., Anal for C₂₆H₂₅ClN₆O₄S: Calculated: C, 56.47; H, 4.56; Cl, 6.41; N, 15.20 found: C, 55.97; H, 4.46; N, 15.20;

AR-52a: 1-(4-(furan-2-yl)-2-(4-(4-methoxyphenyl) piperazin-1-yl) thiazol-5-yl)-2-(2methyl-4-nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 3100 (aromatic stretching), 2829 (-CH₃ stretching), 1630 (C=O stretching), 1566-1536 (C=C stretching, aromatic), 1493-1462 (NO₂-stretching), 1331 (N-C of aromatic), 1152 (C-O-C stretching of morpholine), 1178 (C-O-C stretching of furan), 925-882 (-C=C out of plane for mono substituted benzene)., ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.42 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 3.47-3.76 (m, 8H, piperazine), 3.34 (s, 3H, O-CH₃), 5.46 (s, 2H, -CH2-C=O), 6.66-6.76 (m, 4H, aromatic proton of 4-methoxyphenyl) 7.3-7.42 (m, 3H, aromatic proton of Furoyl at 4^{th} position), 7.88 (s, aromatic proton at 5th position of 1H. nitroimidazole), MS: m/z., 508, 509 (M+1) ., Anal for C₂₄H₂₄N₆O₅S: Calculated: C, 56.68; H, 4.76; N, 16.53 Found: C, 55.68; H, 4.46; N, 16.13;

AR-13a: 2-(2-methyl-4-nitro-1H-imidazol-1-yl)-1-(2-(4-methylpiperazin-1-yl)-4 phenylthiazol-5-

yl) ethanone

IR (KBr, Cm¹): 3151 (aromatic stretching), 2983-2948 (-C-H stretching) 2890-28081 (-CH₃ stretching), 1655 (C=O stretching), 1546 (C=C stretching. aromatic), 1489-1455 $(NO_{2}$ stretching), 1343 (N-C of aromatic), 832 (-C=C out of plane for mono substituted benzene). 1145-1139 (C-N-C stretching of methvl piperazine)., ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.27 (s, 1H –CH₃ Alkane piperazine) 2.44-2.45 (m, 8H, piperazine), 3.33 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 3.49 (s, 3H, O-CH₃), 5.46 (s, 2H, -CH₂-C=O), 7.47-8.18 (m, 5H, aromatic proton at 4th position), 7.13 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 426, 427 (M+1) ., Anal for C₂₀H₂₂N₆O₃S: Calculated: C, 56.32; H, 5.20; N, 19.70; Found: C, 55.22; H, 5.30; N, 18.90;

AR-23a: 1-(4-(4-chlorophenyl)-2-(4methylpiperazin-1-yl) thiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 3141 (aromatic stretching), 2975-2943 (-C-H stretching) 2886-2808 (-CH₃ stretching), 1750-1667 (C=O stretching), 1546 (C=C stretching, aromatic), 1486-1446 (NO₂stretching), 1339 (N-C of aromatic), 890-823 (-C=C out of plane for mono substituted benzene), 1142-1138 (C-N-C stretching of methyl piperazine), 652 (-C-Cl Bending)., ¹H NMR:(400 MHz, DMSO-d6-d6) δ (ppm): 2.27 (s, 1H -CH₃ Alkane piperazine) 2.06-2.59 (m, 8H, piperazine), 3.49 (s, s, 3H proton at 2nd position of nitroimidazole $-CH_3$) 5.43 (s. 2H, -CH₂-C=O), 7.38-8.29 (m, 4H, aromatic proton at 4th position), 7.21 (s, 1H, aromatic proton at 5th position of nitroimidazole)., MS: m/z., 460, 461 (M+1)., Anal for C₂₀H₂₁ClN₆O₃S: C, 52.11; H, 4.59; N, 18.23; Found: C, 52.02; H, 4.29; N, 17.23:

AR-33a: 1-(4-(2-chlorophenyl)-2-(4-methylpiperazin-1-yl)thiazol-5-yl)-2-(2-methyl-4nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 3121 (aromatic stretching), 2975 (-C-H stretching) 2886 (-CH₃ stretching), 1667 (C=O stretching), 1549 (C=C stretching, aromatic), 1486 (NO₂- stretching), 1270 (N-C of aromatic), 890-823 (-C=C out of plane for mono 1142-1138 substituted benzene), (C-N-C stretching of methyl piperazine), 650 (-C-Cl Bending)., ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.15 (s, 1H –CH₃ Alkane piperazine) 2.54 (m, 8H, piperazine N-CH₂), 3.33 (s, 3H proton at 2^{nd} position of nitroimidazole –CH₃) 5.27 (s, 2H, -CH₂-C=O), 7.50-8.01 (m, 4H, aromatic proton at 4th position), 7.13 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 460, 461 (M+1)., Anal for C₂₀H₂₁ClN₆O₃S: Calculated: C, 52.17; H, 4.59; N, 18.23; found: C, 51.11; H, 4.79; N. 18.11

AR-43a: 1-(4-(3-chlorophenyl)-2-(4-methylpiperazin-1-yl)thiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl) ethanone

IR (KBr, Cm¹): 3141-3118 (aromatic stretching), 2952 (-C-H stretching) 2836 (-CH₃ stretching), 1687-1666 (C=O stretching), 1546 (C=C)stretching, aromatic). 1505-1494 $(NO_2$ stretching), 1300-1350 (N-C of aromatic), 807 (-C=C out of plane for mono substituted benzene), 1142-1138 (C-N-C stretching of methyl piperazine), 622 (-C-Cl Bending)., ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.10 (s, 1H –CH₃ Alkane piperazine) 2.42-2.52 (m, 8H, piperazine), 3.35 (s, 3H proton at 2nd position of nitroimidazole –CH₃) 5.30 (s, 2H, –CH₂-C=O), 7.51-8.30 (m, 4H, aromatic proton at 4th position), 7.23 (s, 1H, aromatic proton at 5th position of nitroimidazole), MS: m/z., 460, 461 (M+1)., Anal for C₂₀H₂₁ClN₆O₃S: Calculated: C, 52.11; H, 4.59; N, 18.23; found: C, 52.14; H, 4.49; N, 18.13;

AR-53 a: 1-(4-(furan-2-yl)-2-(4-methylpiperazin-1-yl)thiazol-5-yl)-2-(2-methyl-4-nitro-1H-imidazol-1-yl)ethanone

IR (KBr,Cm¹): 3110 (aromatic stretching), 2978 (-C-H stretching) 2818 (-CH₃ stretching), 1685 (C=O stretching), 1546 (C=C stretching, aromatic), 1489-1455 (NO₂- stretching), 1343 (N-C of aromatic), 1168 (C-O-C stretching of furan), 832 (-C=C out of plane for mono substituted benzene), 1145-1139 (C-N-C stretching of methyl piperazine)., ¹H NMR:(400 MHz, DMSO-d6) δ (ppm): 2.22 (s, 1H –CH₃

Compound	Dose mg/kg	I hr	II hr	III hr
AR-11-a	10	20.19	19.41	5.20
AR-11-a	20	52.68	49.92	35.42
AR-11-a	40	67.51	62.28	48.67
AR-21-a	10	68.77	70.27	64.37
AR-21-a	20	39.43	51.33	52.60
AR-21-a	40	51.10	48.98	49.10
AR-31-a	10	42.27	54.30	61.29
AR-31-a	20	69.72	67.29	68.29
AR-31-a	40	43.22	33.49	38.18
AR-41-a	10	34.38	36.13	-0.106
AR-41-a	20	39.74	28.16	30.01
AR-41-a	40	-15.45	-5.79	-15.9
AR-51-a	10	51.47	56.02	53.86
AR-51-a	20	59.97	48.2	41.24
AR-51-a	40	61.23	55.55	56.62
AR-12-a	20	76.67	75.43	73.17
AR-22-a	20	60.91	35.68	18.34
AR-32-a	20	61.54	58.05	51.11
AR-42-a	20	88.65	87.94	73.38
AR-52-a	20	53.35	58.52	53.65
AR-13-a	20	43.26	64.78	70.62
AR-23-a	20	34.12	-22.92	33.18
AR-43-a	20	76.04	67.91	63.52
Std. Diclofenac	10	77.87	74.64	76.45
Std. Ibuprofen	20	76.24	75.21	74.23

Table 2: Anti-inflammatory activity carrageenan- induced rat hind paw oedema % protection

Oral administration for all test compounds, P < 0.05, Student's t-test versus controls, the standard drugs (dose and % protection) were Diclofenac sodium (10 mg/kg, 75 %) and ibuprofen (20 mg/kg, 74%)

Alkane piperazine) 2.55 (m, 8H, piperazine), 3.32 (s, 3H proton at 2^{nd} position of nitroimidazole –CH₃) 5.46 (s, 2H, –CH₂-C=O), 7.61-8.21 (m, 3H, aromatic proton of furoyl at 4th position), 7.13 (s, 1H, aromatic proton at 5th position of nitroimidazole)., MS: m/z., 416, 417,418 (M+1)., Anal for C₁₈H₂₀N₆O₄S: Calculated: C, 51.91; H, 4.84; N, 20.18; found: C, 51.21; H, 4.44; N, 19.80;

Pharmacological Screening

Animals: Albino rats (150-250 g) of either sex were provided with pellet diet (Lipton, India) and water ad libitum and kept under standard laboratory condition at 25 ± 2°C.The experimental protocol was approved by the institutional ethics committee constituted [RCPIPER/IAEC/2013-14/09]by the Ministry of Social Justice and Empowerment, (Government, of India).

Anti-Inflammatory Activity

We have used the method previously described by Winter et al^{25} . The animals were studied for toxicity of DMSO up to 10% v/v in saline, and 5% DMSO was selected as a vehicle to suspend the standard drugs and the test compounds. Albino rats of either sex weighing between 150 -250 g were starved for 18 hours prior to the experiment. The animals were weighed, marked for identification and divided into groups of six. The standard drug, diclofenac sodium (10 mg/kg body weight) and ibuprofen (20 mg/kg body weight) and the test compounds were given orally (10, 20 and 40 mg/kg body weight) as a suspension using 5% DMSO as a vehicle. One hour later foot paw oedema was induced by injecting mL of 1% carrageenin 0.1 subcutaneously into the planter portion of the right hind paw of each rat. Initial foot paw volume was measured immediately by mercury plethysmometer.

Oedema was measured three hours after carrageenin administration. The swelling in test group animals was used to calculate the percent inhibition \pm SEM of oedema achieved by the compound at the test dose compared with the vehicle control group. The % protection of

oedema was calculated according to the formula, % anti-inflammatory activity = 100 x (1-Vt/Vc)where, Vt and Vc were the volume of oedema in test compounds and control groups, respectively.²⁶

RESULTS AND DISCUSSION

Novel trisubstituted thiazole analogues (AR-11a-AR-53a) were designed and synthesized. The synthesized compounds (AR-11a-AR-53a) were evaluated for their in vivo anti-inflammatory activity. The in vivo anti-inflammatory activity was performed in carrageenin-induced rat hind paw oedema model at three graded doses employed at 10, 20 and 40 mg/kg body weight using diclofenac sodium and Ibuprofen. The chemical structures of these compounds results of in vivo anti-inflammatory activity are shown in table 1 and 2 respectively. Taking into account the assorted biological activities of (1-chloro-3-(2-methyl-4-nitro1H-imidazole-1-yl) propane-2derivatives as important member of on antiprotozoal agent was used at fifth position of thiazole, However it is evident that antiinflammatory activities of thiazole derivatives are extensively reported and clinically proved. The compounds (AR-11a-AR-53a) were designed and synthesized by keeping (1-chloro-3-(2-methyl-4nitro1H-imidazole-1-yl) propane-2-on at fifth and morpholin/ position methoxyphenylpiperazinyl / methylpiperazinyl moiety at second position of thiazole ring. The forth position of thiazole was substituted by introducing electron withdrawing group (-Cl), at different position in phenyl moiety.

The compounds (*AR-12a to AR-53a*) were synthesized having methoxy and methyl group at para position of phenylpiperazinyl and piperazinyl respectively to explore the effect of presence of substituted aromatic ring and aliphatic chain on biological activity of profile of the candidate.

The compounds (*AR-11a to AR-51a*) were synthesized having morpholin ring at second position of thiazole. Among all the tested compounds *AR-21a*, *AR-31a*, *AR-12a* and *AR-42a* showed maximum anti-inflammatory activity of 70% protection at 20 mg/kg and 56% protection at 40 mg/kg to inflamed paw, while other compounds showed Moderate to less protection at 10 mg/kg, 20 mg/kg&40 mg/kg to inflamed paw.

In the series the first compound from each set were screened at three graded doses employed at 10, 20 and 40 mg/kg body weight using diclofenac sodium. While rest of the compounds were screened at single grade dose of 20mg/kg

In the series, AR-21a showed comparable antiinflammatory activity at all the three graded doses employed, i.e. 64.76% at 10 mg/kg, 52.59% at 20 mg/kg and 49.09% at 40 mg/kg, for AR-31a the % protection was found to be 61.29% at 10 mg/kg, 68.29% at 20 mg/kg and 38.16% at 40 mg/kg, Similarly for AR-12a and AR-42a the % protection was displayed 73.17% and 73.38% at 20 mg/kg respectively which are the highest % protection among the mentioned series.

Other Compounds showed moderate to less activity as compared to above four compounds. All the targeted trisubstituted thiazole (AR-11a to AR-53a) were found to have moderate antiinflammatory activity as compared to the standard drug diclofenac sodium and Ibuprofen, which showed 75% protection at 10 mg/kg dose and 74% protection at 20 mg/kg dose.

To summarize these findings, methyl substituent on piperazine moiety and *p*-chlorophenyl at second and forth position of thiazole respectively enhances lipophilicity & potency of tested compounds.

The presence of methoxyphenylpiperazinyl / methylpiperazinyl moiety and chlorophenyl at second and forth position of thiazole respectively enhances the anti-inflammatory activity. However all targeted trisubstituted thiazole containing (1-chloro-3-(2-methyl-4-nitro1Himidazole-1-yl) propane-2-on were found to have good to moderate anti-inflammatory activity, it is therefore, suggested that this scheme to be studied further to explore its full structureactivity relationship potential particularly single candidate as monotherapy in the treatment of chronic inflammatory diseases having both antiinflammatory and antimicrobial activities.

CONCLUSION

The synthesized targeted compounds (*AR-11a to AR-53a*) were evaluated for their *in vivo* antiinflammatory activity. On the basis of structureactivity relationship studies of (*AR-11a to AR-53a*) it can be concluded that presence of (–Cl) chloro substituted electron withdrawing group at different position in phenyl moiety at the fourth position of the thiazole contributes significantly to anti-inflammatory activity profile of the candidates, similarly by keeping (1-chloro-3-(2methyl-4-nitro1H-imidazole-1-yl) propane-2-on at fifth position it is predictable to have good anti-infective property which highly desirable to be evaluated.

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